### Chikkanna Government Arts College, Tirupur-641 602

2.5.1 - Mechanism of internal assessment is transparent and robust in terms of frequency and mode. Write description within 200 words.

The internal assessment is made in accordance to the guidlines of the parent University. However, the college ensures at most transparency in the conduct and evaluation of continuous internal tests and award of internal marks to students. Measures taken to ensure transparency in internal assessment mechanisms are given below:

- 1. Conduct of Internal Assessment Tests three times in a semester in accordance to the schedule given in the academic calendar
- 2. Evaluation of Answer Scripts and Assignments and disclosing the score to the students immediately
- 3. Record of Internal Assessment Marks in CIA Register: CIA Marks for each subject under each parameter is recorded in a separate register. This register is maintained separately for each batch of students and all CIA marks are entered for their entire course
- 4. Internal marks are disclosed to the students and discussed with them. The final marks after improvement if any is recorded in the CIA registers and students signatures are obtained after their agreement on their scores.
- 5. Periodically, the internal assessment scores are intimated to the parents

### **Internal Exam Attendance**

			Chikkanna Government A	rts College, Tirupur-2		
	3		PG Department of	of Chemistry		
			I CIA Exam F			
Semes				III B.Sc(T/M)		
S.No	Date		27-03-202		23 27-03-20	22/
	Subject		Organic-II 63A	Physical-63B	Dye Chem-6EC	23 02-03-202 EVS-6EE
	1 2022D0025	AISWARYA L	+A3-	5-BAUNDI L. FTISHO		
	2 2022D0026	BRINDHA S	-AB-	S. Brindha	S. Brindha	1. Alswarya
	3 202200027	KIRUBASRI D	DKirubashi	D. Kirubalen	20	D. Boundha
	4 2022D0028	KULANTHAI THERAS A	Kulaerthour therain	A Kulastitai Atrora	The works	D. Kirubasy
	5 2022D0029	UMA R	R. Oma	R. Uma.		and kulonthai thoray
	6 2022D0030	ARULKUMAR K	(1). Jor Blong	(b). spriBlom.	B. Omer.	R. DMA
	7 2022D0031	DHARUN R	R. Dharus	B. Dharus	100	10/20m/3 6m
	S 2022D0032	DRAVIDAN J	Pag Sooi	Detai	P. pharum	Resharing 5
	9 2022D0033	KALAISELVAN K	Valairelvanu	water bluran le	Vi Cong	The state of the s
	10 <b>2022D0034</b>	KARTHICK G	Karthick G	Karthick-G	Karthiele. Co	Kalaulelian.u
1	11 2022D0035	KARTHICK M	(4) Haringalio	(1) · Francisco		excettick-co
	12 2022D0036	MOHAMMEDSHARIF A	dit A	that A	A LA	6) prince
	13 2022D0037	PASUPATHI M	H. PasuPath.	M. BLUDITA	M-ParoBHi	M. Dag Patti
_	14 2022D0038	PRADEEP V	Y . Para dap.	V. ParadelP.	v-PradeeP.	V. Ponderp.
	15 2022D0039	PRAVEEN KUMAR S	5. Prayoen kumar	5. Praveenkumar	S. Praveen kumar	S. Prayoon kumar
1	15 2022D0040	PRAVEEN S	PRAVEEN.S	PRAVERN-S	PRAVERIN.S	PRAVEEN.3
	2022D0041	RAGUL M	Round.M	Roand.M	Road . w.	Ragul. M
1	2022D0042	RANJITHKUMAR N	Ranthaken. N	Dayoth Ling. N	Bayes Kup N	12.000
1	.9 <b>2022D0043</b>	SANTHOSH KUMAR S	S. South a	5. 2. 141 - T	S. L. bett. o	So Southed have
2	0 2022D0044	SELVAKUMAR A	A. Selvakumas	A. selvaicened	A. Selvalumos	A. Solvakust
2	2022D0045	SIVA M	M. 9772.M	M. CIVA.	M.SIVA	M. SIVA 1
- 2	2 2022D0046	SRIJEGADEESH R	R. Soutegados		R. Sui Jegardes .	
2	3 2022D0047	THAMIMANSARI A	A. Thanin Annari	A Thanim Array	A Thanim Away	R-Svitegerdech.





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			I CIA	Exam Feb 2023			
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S.No	-	Subject		Organic-II 63A	Physical-63B	Dye Chem-6EC	EVS-6EE
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	7	2022D0007	ANANTHAN R	R. Aventhour	P. Franthan	R. franthan	R Tray Than
	8	2022D0008	ANBARASAN M	M. Arlm	mediant	ey. allum	M. on hom
	9	2022D0009	DEEPAK KULANTHAI RAJ S R	Ruft.	Reyport.	Reepst.	(ARSENT)
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	11	2022D0012	ELAMURUGAN S	S. Duca.		S-Edyane.	3. Elya.
	12	2022D0013	GAJENDRAN S	8. Gajendran	S. Gajardren	3. Gojerchan-	9 Gajardran.
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### Sample Internal Exam Timetable

	Sample Internal Exam Timetable	
	PG & Research Department of Chemistry	
Contin	Leternal Aggreement (CIA II) Test - MARCH 2023	
	& M.Sc Chemistry  Time: 9:30 AM to 11:30 AM	
0.30	Subjects	Subject Code
Date	Subjects	22T
7.03.23	Tamil - II	42T
	Tamil - IV	63A
	Core-XII Chemistry Paper X (Spectroscopy & Natural Products)	23A
M.Sc	Organic Chemistry II ( Molecular Rearrangements & Photochemistry)	
	Inorganic Chemistry II (Coordination Chemistry)	43A
		22E
28.03.23	English - IV	42E
	Core XIII Chemistry Paper XI (Physical Chemistry)	63B
M C-	Physical Chemistry II (Quantum Chemistry & Nanomaterials)	23B
M.50	Physical Chemistry IV (Reaction Kinetics and Electrochemistry)	43B
29.03.23	Allied Physics	2AF
29.03.23	Allied Mathematics and Botony	4AA & 4A
*	Elective Group II C - Dye Chemistry	6EC
M.Sc	Physical Methods in Chemistry I - Paper VI	23C
11115	Polymer Technology - Paper XII	43C
30.03.23	Core II - Chemistry Paper II (General Chemistry II)	23A
20102	Core VI - Chemistry Paper V - Organic Chemistry I	43A
	Elective Group III B: Environmental Chemistry	6EE
M.S	Elective II: Water Pollution and Industrial Effluent Treatment	2EA
	Elective IV: Industrial Chemistry	4EA
31.03.23	Skill Based Subject: Technology of Dyeing of Natural Fiber(Tex.Chem II)	4ZB
01.04.23	Allied Chemistry (for Zoo, Maths and Physics Major)	2AH

Head

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### **Internal and External Mark Record - Sample**

# CHIKKANNA GOVERNMENT ARTS STUDENTS RECORD

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## COLLEGE, TIRUPUR - 641 602.

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# CHIKKANNA GOVERNMENT ARTS STUDENTS RECORD

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### COLLEGE, TIRUPUR - 641 602. SHEET

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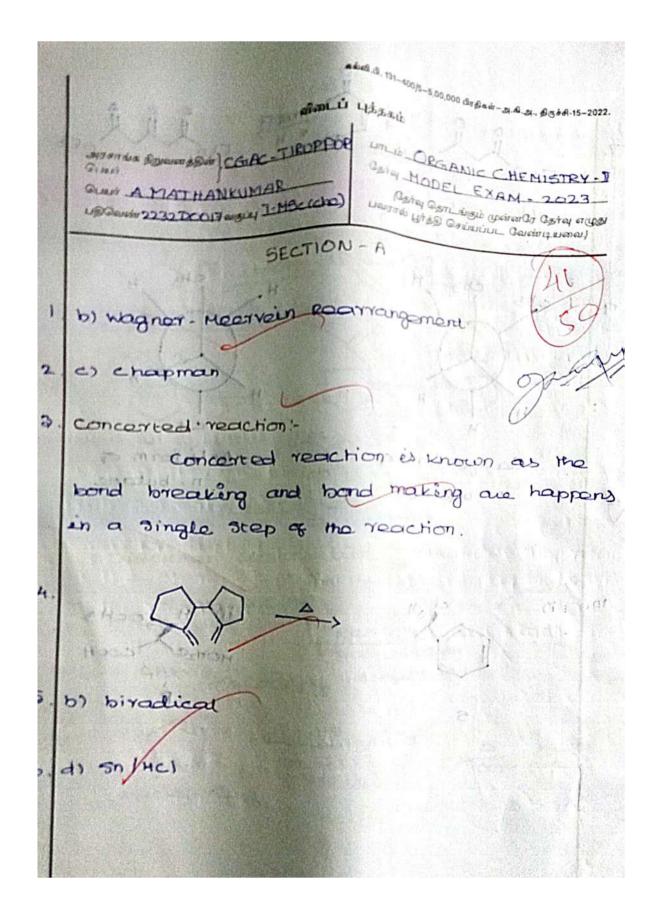
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3) Course Completed / Discontinued

4) TC lasued 5) Degree Cartificate Issued

1) Free Bus Pass 2) Free Laptop

### Sample Internal Exam Paper



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### Sample Assignment

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II. M.Sc Chemisty

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ii) calculate the wave number of the absorption peak for Del assuming the force constant is the same as that obtained in (i)

$$\bar{V} = \frac{1}{2\pi c} \int \frac{K}{\mu}$$

$$= \frac{1}{2 \times 3.14 \times 3 \times 10^{10} \text{ cm/s}} \sqrt{\frac{480.30 \text{ kgs}^2}{2 \text{ glmol} \times 35.5} \text{ glno}}$$

$$= \frac{1}{2 \times 3.14 \times 3 \times 10^{10} \text{ cm/s}} \sqrt{\frac{2 \text{ glmol} \times 35.5}{2 \text{ glno}} \text{ glno}}$$

ASSIGNMENT-I

N. SENTHILVEL

II MSC. CHEMISTRY

REGI. NO: 2232D0019

PHYSICAL METHODS IN CHEMISTRY-II

DATE: 23 |08/203

Calculation of Amon of Amonatic system using Woodward Fisher Rule i) Base value for a) Arcor - 216 nm b) Archo - 250 nm c) Accort - 230 nm d) Arwar - 230 nm ii) Alkyl geoup of Ring Residue in Ortho And Meta position - 3nm iii) Alkyl geoup of Ring Residue in pare position - wom. iv) polar geoups 1a) - oH, - oCH3; oAley in o, m Pasition - 7 nm b) -OH, -OCH3, -OALKy P Position - 25 nm - 11 nm c) = 0 (oxonium) in o position. - 20 nm d) -o (oxonium) in m position E) -d in o, m position - 0 nm g) -c1 in p position - 10 nm A) - Br in o, m position - 2 nm h) - 0 (oxorium) in p position - 78 nm i) -Br in p position - 15 nm j) -NH2 is 0, m position - 13 nm K) - NH2 in p position. - 58 nm 1) -NHIOCH3 in o.m position - 20 nm

A5 nm

m) - NH LOCH3 in p position